

5-Aminopyrimidine

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Key indicators

Single-crystal X-ray study
 $T = 298\text{ K}$
 Mean $\sigma(\text{C}-\text{C}) = 0.001\text{ \AA}$
 R factor = 0.039
 wR factor = 0.130
 Data-to-parameter ratio = 17.3

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

In the title compound, $\text{C}_4\text{H}_5\text{N}_3$, the nearly planar 5-amino-pyrimidine molecule has twofold rotation symmetry. The crystal structure is stabilized by a hydrogen bond between the amino group and the ring N atoms, thus forming a two-dimensional network parallel to the ab plane.

Comment

Recently, there has been considerable interest in the use of pyrimidine (pym) as a bridging ligand for the formation of coordination polymers. Strong magnetic couplings can be mediated by μ -bonded pym ligands. One-, two- and three-dimensional structural motifs have been studied. For example, $\text{Cu}(\text{NO}_3)_2(\text{pym})(\text{H}_2\text{O})_2$ behaves as a uniform $S = \frac{1}{2}$ anti-ferromagnetic chain (Feyerherm *et al.*, 2000; Yasui *et al.*, 2001). Structurally, the complex $\text{Cu}(\text{dca})(\text{NO}_3)(\text{pym})(\text{H}_2\text{O})$ (dca = dicyanamide) is two-dimensional, but magnetically it behaves as a one-dimensional chain, because the magnetic coupling through the 3-atom pym bridges is significantly stronger than that through the 5-atom dca bridges (Manson *et al.*, 2003). Examples of structurally three-dimensional materials include $\text{Cu}_3(\text{dca})_6(\text{pym})_2 \cdot 0.75\text{H}_2\text{O}$ (Manson *et al.*, 2003) and $\text{Cu}(\text{HCO}_2)_2(\text{pym})$ (Manson *et al.*, 2005). The former of these has a large exchange coupling constant ($J/k_B = -69.4\text{ K}$), while the latter exhibits long-range magnetic ordering below $T_N = 2.8\text{ K}$. Spontaneous magnetization is also observed in the three-dimensional complexes $M(\text{dca})_2(\text{pym})$ ($M = \text{Fe}$ or Co), with ordering temperatures of 3.2 and 1.8 K, respectively (Kusaka *et al.*, 2000).



We are interested in increasing the dimensionality in these systems through the use of pyrimidine derivatives with hydrogen-bonding functionalities. Aminopyrimidines (apym) are one promising family that has this capability. 2-Aminopyrimidine (2-apym) has been used extensively as a ligand in coordination complexes. For example, a novel molecular tube-like structure containing monodentate 2-apym ligands and the relatively rare $\mu_{1,3,5}$ coordination mode for the dca anions has been reported for $M(\text{dca})_2(2\text{-apym})$ ($M = \text{Co}$ or Ni) (Jensen *et al.*, 2000). Monodentate 2-apym ligands are also found in the complex $\text{Cu}(\text{dca})_2(2\text{-apym})_2$, which forms one-dimensional dibridged $\mu_{1,5}$ -dca chains (van Albada *et al.*, 2000). When 2-apym acts as a bridging ligand, strong exchange coupling

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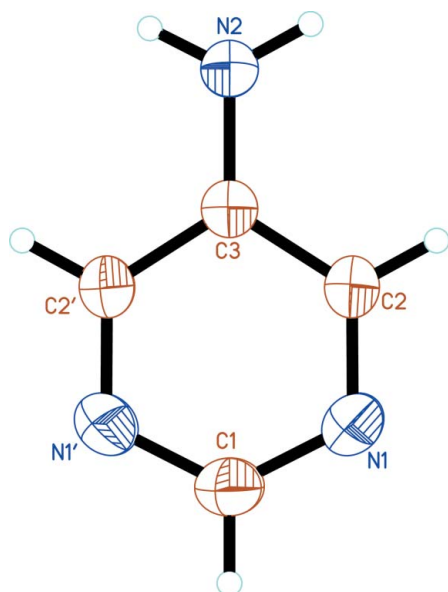


Figure 1

A view of the molecular structure of 5-aminopyrimidine, showing the atom-numbering scheme. The prime corresponds to symmetry code (i) in Table 1. Displacement ellipsoids are drawn at the 50% probability level and H atoms are shown as small spheres of arbitrary radii.

contants can be obtained, *e.g.* in $[\text{Cu}_4(2\text{-apym})_6(\mu\text{-OCH}_3)_2(\mu\text{-F})_3(\text{F})_2](\text{BF}_4)$ ($J/k_B = -274$ K) (van Albada *et al.*, 2003). Another example of bridging 2-apym ligands is found in the zigzag chain structure of $\text{Cu}_2(\text{acetate})_4(2\text{-apym})$ (Smith *et al.*, 1991; Blake *et al.*, 2002). We are aware of no coordination complexes derived from 4-apym or 5-apym. The N atoms of the 2-apym and 4-apym derivatives are more sterically hindered than the 5-apym derivative, thus making 5-apym a promising candidate for a bridging ligand. While the crystal structure of the 4-apym ligand has been published (Van Meervelt & Uytterhoeven, 2003), we report here, for the first time, that of 5-apym, *viz.* (I).

The 5-apym molecule lies on a twofold rotation axis. The atom-numbering scheme is shown in Fig. 1. The bond lengths and angles are typical of pyrimidines, including pym (Wheatley, 1960; Furberg *et al.*, 1979), 2-apym (Scheinbeim & Schempp, 1976; Furberg *et al.*, 1979) and 4-apym (Van Meervelt & Uytterhoeven, 2003). The ring atoms deviate only slightly from coplanarity, with atoms N1 and C2 out of the least-squares plane by 0.0022 (6) Å. By symmetry, the amino N atom lies in the plane of the pym ring. The dihedral angle between the plane of the amino group and the least-squares plane of the ring is 9.4 (17)°, significantly smaller than the 22° angle observed in 2-apym (Scheinbeim & Schempp, 1976).

As illustrated in Fig. 2, the packing of the 5-apym molecules is stabilized by an $\text{N2} \cdots \text{H3} \cdots \text{N1}^{\text{i}}$ hydrogen bond (Table 2). The hydrogen-bond network results in the formation of two-dimensional sheets parallel to the *ab* plane. The pym plane is tilted by 11.50 (6)° with respect to the *ab* plane. Adjacent sheets are arranged such that uniform slipped stacks of 5-apym molecules form along the *a* + *b* diagonal. The 5-apym centroid-centroid distance of 3.723 Å and perpendicular

separation of 3.362 Å confirm the presence of π - π interactions (Spek, 2003).

Experimental

5-Aminopyrimidine was prepared according to the literature procedure of Phillips *et al.* (1999). Single crystals suitable for X-ray diffraction were grown by recrystallization from benzene.

Crystal data

$\text{C}_4\text{H}_5\text{N}_3$
 $M_r = 95.11$
 Monoclinic, $C2/c$
 $a = 7.5982$ (10) Å
 $b = 10.1226$ (13) Å
 $c = 6.8106$ (10) Å
 $\beta = 118.596$ (5)°
 $V = 459.93$ (11) Å³
 $Z = 4$

$D_x = 1.374$ Mg m⁻³
 Mo $K\alpha$ radiation
 Cell parameters from 1724 reflections
 $\theta = 3.4\text{--}29.5^\circ$
 $\mu = 0.09$ mm⁻¹
 $T = 298$ (2) K
 Rod, colourless
 $0.50 \times 0.22 \times 0.20$ mm

Data collection

Siemens SMART CCD area-detector diffractometer
 ω scans
 Absorption correction: integration (*XPREP* in *SHELXTL*; Sheldrick, 2001)
 $T_{\min} = 0.956$, $T_{\max} = 0.987$
 2534 measured reflections

641 independent reflections
 584 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.023$
 $\theta_{\max} = 29.5^\circ$
 $h = -10 \rightarrow 10$
 $k = -14 \rightarrow 13$
 $l = -9 \rightarrow 9$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.039$
 $wR(F^2) = 0.130$
 $S = 1.12$
 641 reflections
 37 parameters
 H atoms treated by a mixture of independent and constrained refinement

$w = 1/[\sigma^2(F_o^2) + (0.0723P)^2 + 0.0952P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} = 0.007$
 $\Delta\rho_{\max} = 0.31$ e Å⁻³
 $\Delta\rho_{\min} = -0.17$ e Å⁻³

Table 1

Selected geometric parameters (Å, °).

| | | | |
|--------------------------|-------------|---------------------------|-------------|
| C1—N1 | 1.3298 (11) | C2—C3 | 1.3993 (11) |
| C2—N1 | 1.3262 (14) | C3—N2 | 1.3531 (17) |
| N1—C1—N1 ⁱ | 125.97 (14) | C2—C3—C2 ⁱ | 114.65 (12) |
| N1—C2—C3 | 122.97 (9) | C2—N1—C1 | 116.72 (9) |
| N2—C3—C2 | 122.68 (6) | C3—N2—H3 | 117.7 (12) |
| N1—C2—C3—N2 | 179.77 (6) | C3—C2—N1—C1 | 0.43 (12) |
| N1—C2—C3—C2 ⁱ | −0.23 (6) | N1 ⁱ —C1—N1—C2 | −0.21 (6) |

Symmetry code: (i) $-x, y, -z + \frac{1}{2}$.

Table 2

Hydrogen-bond geometry (Å, °).

| $D\cdots H\cdots A$ | $D\cdots H$ | $H\cdots A$ | $D\cdots A$ | $D-H\cdots A$ |
|---|-------------|-------------|-------------|---------------|
| $\text{N2} \cdots \text{H3} \cdots \text{N1}^{\text{ii}}$ | 0.89 (2) | 2.23 (2) | 3.1078 (11) | 170 (2) |

Symmetry code: (ii) $-x + \frac{1}{2}, y + \frac{1}{2}, -z + \frac{1}{2}$.

H atoms on aromatic C atoms were positioned geometrically and refined with a riding model, with $\text{C}—\text{H} = 0.93$ Å. The amino-group H atom was located in a difference map and its position freely refined. For all H atoms, $U_{\text{iso}}(\text{H})$ was constrained to be 1.2 (aromatic) or 1.5 (amino) times U_{eq} of the carrier atom.

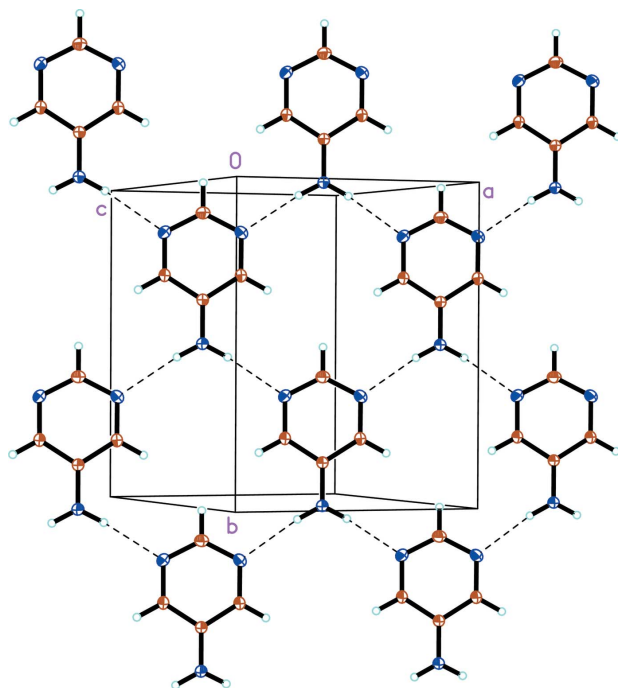


Figure 2

A packing diagram for 5-aminopyrimidine, illustrating the two-dimensional network in the *ab* plane. Displacement ellipsoids are drawn at the 20% probability level and H atoms are shown as small spheres of arbitrary radii. Hydrogen bonds are depicted as dashed lines.

Data collection: *SMART* (Bruker, 1997); cell refinement: *SAINT* (Bruker, 2001); data reduction: *SAINT*; program(s) used to solve structure: *SHELXTL* (Sheldrick, 2001); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL* and *PLATON* (Spek, 2003).

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